

Good welding practice Stainless Steels

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Stainless Steels

- Five basic types of stainless steels,
 - Austenitic, most common
 - Ferritic
 - Martensitic
 - Duplex, main use oil & gas
 - Precipitation Hardened
- All have different properties and hence different applications:
- Most are weldable, but with different problems.
 - Special case Precipitation Hardened



Typical Stainless applications

- Austenitic
 - Good corrosion resistance, low temperature ductility, high temperature oxidation resistance
 - process plants, cryogenics, corrosive applications, used over wide temperature range
- Ferritic
 - Lower cost variant, good strength but lower corrosion resistance
 - bulk handling equipment, road/rail vehicles, not for elevated temperatures (<300°C)

Martensitic-

- High strength/high hardness,
- aerospace, medical, petrochemical

Duplex-

- Good strength and corrosion resistance
- oil & gas extraction/production, not for elevated temperatures (<300°C)



Arc Welding Processes

ARC WELDING

MANUAL METAL ARC
TUNGSTEN INERT GAS SHIELDED
METAL ACTIVE GAS SHIELDED
FLUX CORED ARC WELDING
SUBMERGED ARC WELDING
PLASMA ARC WELDING
ELECTRON BEAM WELDING
LASER BEAM WELDING

SOLID PHASE WELDING

FRICTION WELDING	FW
FRICTION STIR WELDING	FSW

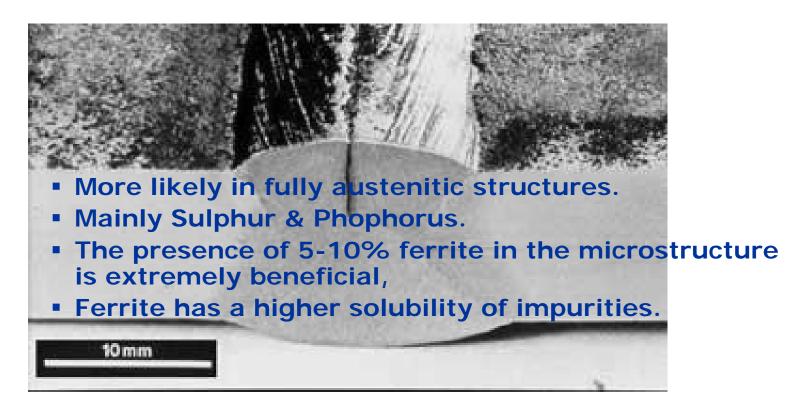
MMA(SMAW)TIG(GTAW)MAG(GMAW)FCAWSAWSAW-PAW-EBW-LBW-



Austenitic stainless steels Welding Problems



- Welding problems
 - Weld metal solidification cracking





Solidification cracking prevention.

Welding Consumables

- The choice of filler material composition is crucial.
- Filler choice depends on required mechanical and corrosion properties – often higher alloy (nominally matching)
- Autogenous welds are possible, but usually welded with matching filler or slightly overalloyed.
- e.g. when welding Type 304 stainless steel, a Type 308 filler material which has a slightly different alloy content, is used.
- Weld metal composition can be predicted from the Schaeffler diagram



Schaeffler diagram

Stainless steel composition determines whether solidification is initially ferrite (δ), austenite (γ), or a mixture of both (δ + γ)

Austenite stabilisers, Ni_{eq}

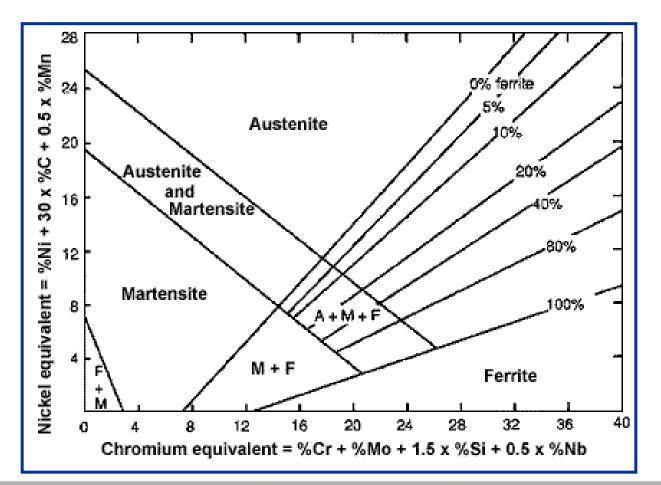
 Carbon, Nickel, Nitrogen, Manganese, Copper Ferrite stabilisers, Cr_{eq}

 Chromium, Molybdenum, Tungsten, Niobium, Silicon

Ferrite measured by magnetic instrument or by metallography



Schaeffler diagram predicts weld metal phases

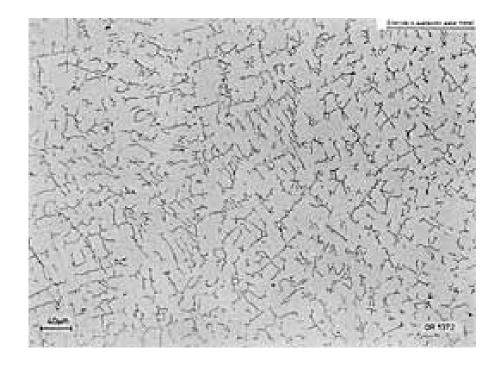




Weld microstructure

Weld bead

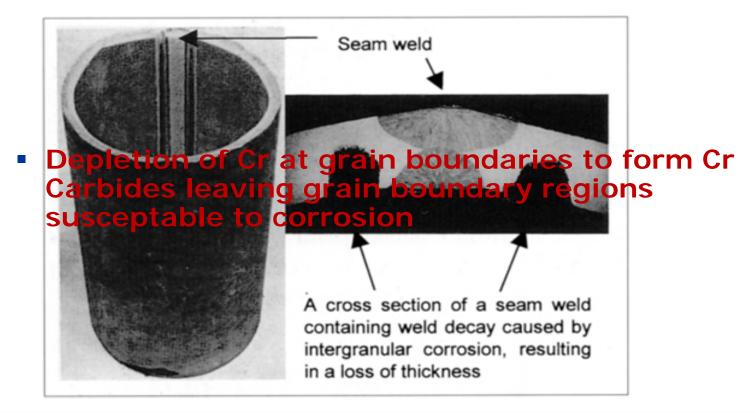
Delta ferrite (dark regions)



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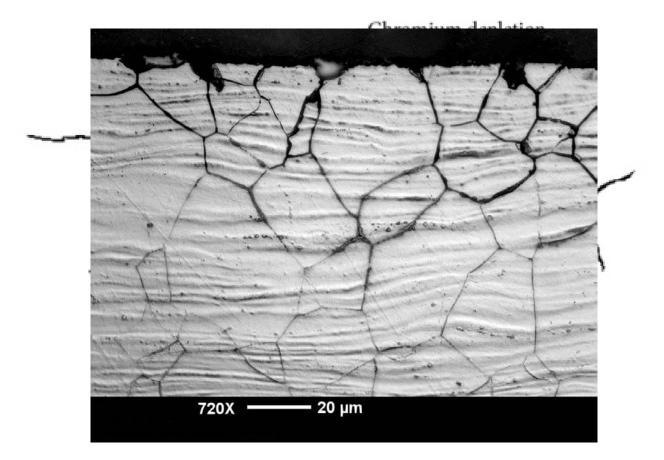


- Welding problems
 - Weld Decay (sensitisation)





Weld Decay (sensitisation)



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- Weld Decay (sensitisation) prevention.
 - Materials
 - Use stabilised materials, grades 321 (Ti) or 347 (Nb).
 - Ti or Nb carbides form in preference to Cr leaving Cr in grains for protection.
 - Use L grades with <0.03% Carbon,</p>
 - insufficient C to form carbides



- Welding issues;
- Distortion
 - Low Thermal conductivity
 - 16 W/m°C compared to 47 W/m°C for Csteel
 - High Coefficient of Expansion
 - 20 µm/m°C compared to 13 µm/m°C for Csteel
 - Minimise
 - Weld volume
 - Accurate fit up:
 - avoid wide gaps and misaligned edges
 - Maintain low interpass temperature (~150°C max).
 - Avoid high heat input techniques
 - i.e. large weave, high welding current, slow travel speed



- Other issues:
 - Weld cold: no problem with phase transformations and avoids precipitation
 - No preheat required, as cold cracking not an issue.
 - Austenite solubility of H₂ is 8x that of Ferrite
 - Rutile electrodes are acceptable and easier to use.
 - Avoid Post-weld heat treatment,
 - not normally required
 - risk of precipitation and sensitisation, 500 -800°C
 - Avoid non-removable backing rings
 - risk of crevice corrosion



Ferritic stainless steels Welding Problems



- Sensitive to grain growth
 - Ioss of toughness, embrittlement
 - Imited to 250°C service temperature
 - can lead to cracking in weld or HAZ of highly restrained joints and thick section material.
- Hydrogen cracking
 - Can suffer cold cracking similar to Carbon Steels.
- Filler metals:
 - matching, austenitic or nickel alloy



Ferritic stainless steels

Usually parent materials are thin plates or tubes.

Austenitic Fillers

- When welding thin section material, (< 6mm) use an austenitic filler and no special precautions are necessary.
- In thicker material, if using an austenitic filler to produce a tougher weld metal, it is necessary to employ a low heat input and max interpass of 150 °C to minimise the width of the grain coarsened zone.
- Nickel Fillers
 - Used if PWHT required



Ferritic stainless steels

Matching Fillers

- If welded with matching filler, preheat and PWHT (due to hard martensitic formation).
- Preheat temperature should be within the range 50-250 °C depending on material composition.
- Preheating will reduce the HAZ cooling rate, avoiding formation of martensite and reduce residual stresses.



Martensitic stainless steels Welding Problems



Martensitic stainless steels

- Welding Problems
 - Hydrogen cracking
 - Normally matching consumables.
 - High hardness in the HAZ makes this type of stainless steel very prone to hydrogen cracking above >3mm. The risk of cracking increasing with the carbon content.

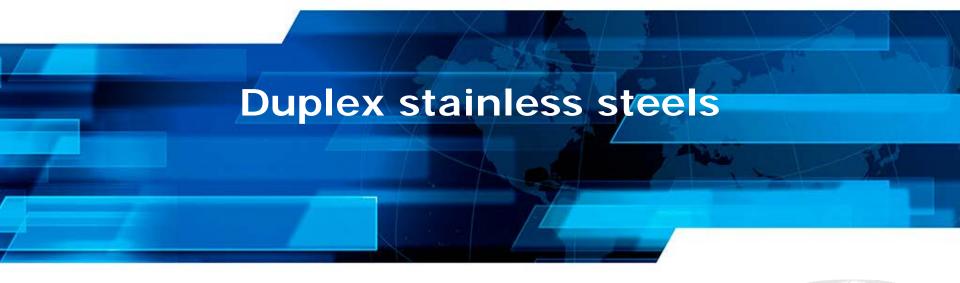
Take precautions to minimise hydrogen



Martensitic stainless steels

- Precautions which must be taken to minimise the risk, include:
 - using low hydrogen processes
 - preheating to around 200 to 300 deg.C.
 - maintaining the recommended minimum interpass temperature.
 - Allow cooling to ambient to ensure complete transformation to martensite prior to PWHT
 - carrying out post-weld heat treatment, e.g. at 650-750 deg.C.

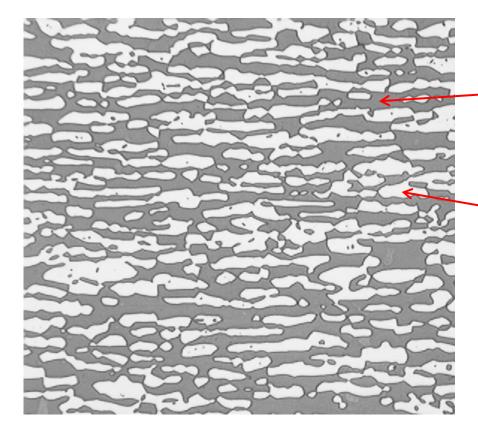






Duplex stainless steels

Complex !



Ferrite matrix - dark High strength Resistance to SCC

Austenite - light Corrosion resistance Toughness

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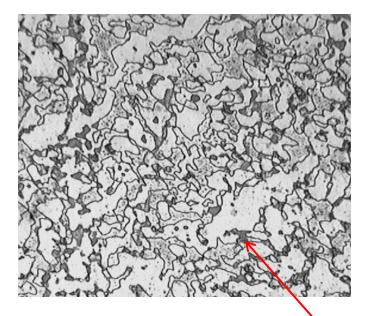
Duplex stainless steels

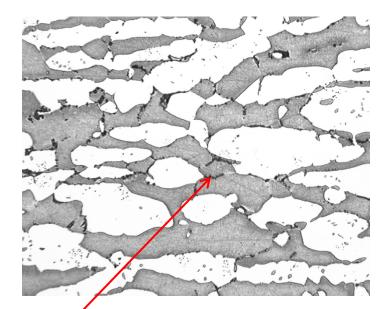
- Complex !
- Preheat is not normally required.
- Heat input and the maximum interpass temperature must be controlled.
 - Too high cooling rate produces high % Ferrite
 - Too slow cooling rate precipitates third phases
- Choice of filler
 - critical to produce a weld metal structure with a ferrite-austenite balance to match the parent metal requirements in as welded condition.
 - To compensate for nitrogen loss, the filler may be overalloyed with nitrogen or the shielding gas itself may contain a small amount of nitrogen.



Duplex stainless steels

Complex ! Get it wrong.





Third phase precipitates Low toughness Low corrosion resistance

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Welding Stainless steels to Carbon/Low alloy steels



- Consumables
 - Use Austenitic consumables or consumables matching stainless grade, alternatively use Ni based consumables.

TWI Stainless to Carbon/Low alloy steels

• Austenitic Stainless

- Welding Problems
 - Dilution of weld metal from the Low alloy base material .
 - possible formation of hard brittle structures,
 - use an overalloyed austenitic stainless filler material, type 309 (24% Cr 12% Ni), weld deposit will be ductile & can tolerate high dilution from carbon /low alloy steel
 - pre-heat is not usually required
 - D Upto 15% ferrite to avoid hot cracking
 - Not suitable for PWHT or above 400°C due sigma phase formation.

TWI Stainless to Carbon/Low alloy steels

- Austenitic Stainless
 - Welding Problems
 - Alternative Ni-based consumables (ENiCrFe-2 and ENiCrFe-3, Alloy 600 type) for PWHT.
 - **PWHT still a problem for the Austenitic Stainless**
 - **Recommended fabrication sequence for PWHT.**
 - 1. Butter the ferritic steel with Ni-base consumables
 - 2. Apply the PWHT to ferritic steel+buttering
 - 3. Complete the joint with Ni-based consumables



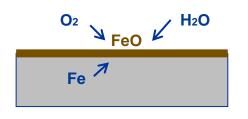
Cross contamination presentation

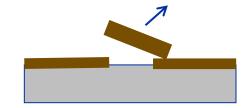


Carbon/Low alloy steels

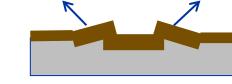
• $Fe + O_2 = Fe_xO_y$ (Rust)

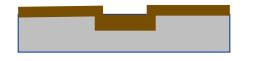
 Requires an electrolyte for the reaction to occur, normally water, moisture in the atmosphere can be sufficient.





But FeO is reactive & 4x volume of Fe. Stresses created in formation cause FeO to flake off. Exposing more Fe.





Which in turn corrodes forming more FeO

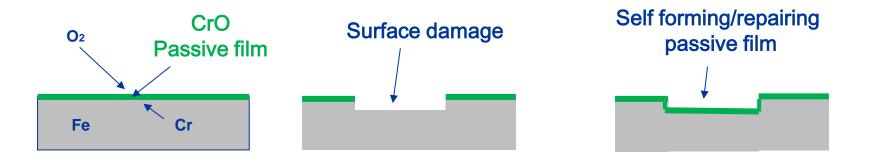
And the process continues, reducing steel thickness.





Basic Corrosion

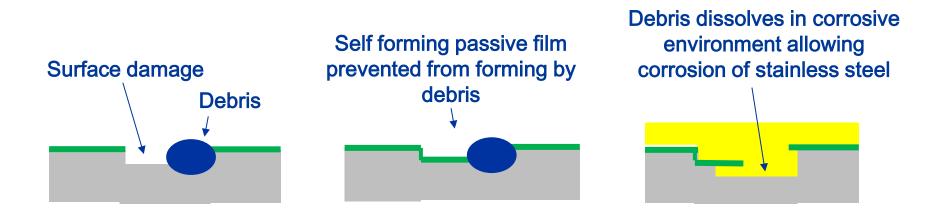
- Stainless steels contain more than 11% Chromium.
 - Cr is more reactive with O₂ than Fe and consequently a Cr oxide is formed in preference to Fe_xO_Y and forms without an electrolyte.
 - Cr_xO_Y is a coherent passive layer (protective film), which is approx. 10 micron thick.
 - Normally self repairing in air, within seconds.
 - The CrO layer prevents further corrosion of the steel.





Basic Corrosion

If Stainless steel is damaged with non corrosion resistant material, i.e. Carbon Steel Passive layer is prevented from reforming This can then lead to corrosion of stainless steel in a corrosive environment.







- Prevent carry over of particulate contamination
 - avoid contact with carbon steels

 chains, forks, benches and hammers

 consider forming processes

 Rolls, pressbrakes and guillotines
 use dedicated tools for stainless steel
 INOX grinding discs, SS wire brushes
 avoid walking on stainless steel material
 Grinding Dust on soles of boots



Fabrication.

- Clean joint and adjacent parent material to remove dirt, grease, oil, paint.
 - Heat from welding can cause elements in these to attack CrO layer or diffuse into microstructure, grain boundaries.
- Temporary attachments
 - materials should be compatible with the parent material
 - preventing dilution and contamination



Fabrication

Arc strikes

damage the CrO layer , creating oxidation act as initiation points for pitting corrosion and cracks

Other considerations

use chloride free marker pens, tapes etc. Inks & glues may contain halides which will attack the CrO layer.



Detection.

 Detection of free iron contamination on the surface by Ferroxyl test, only applicable to austenitic's. Contains Potassium ferricyanide

Post cleaning.

 Appropriate post-weld cleaning to obtain good corrosion resistance, remove spatter, slag, arc strikes, oxides (discolouration) around weld.

Pickle and passivate

- **if required to restore protection.**
- Note. Only restores surface not contamination within the weld metal.



Stainless steels corrode!

- Welds are more prone to corrode than parent material of same composition.
- Intergranular corrosion (weld decay)
- Crevice corrosion
- Ditting corrosion
- Stress Corrosion cracking
 - Chloride Stress Corrosion cracking
 - Sulphide Stress Corrosion cracking
- Cleanliness is important to avoid initiating corrosion



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End of presentation Any Questions glenn.allen@twi.co.uk